Please note the following updates and corrections for Master the PCAT. The corrections indicated below are made when the book is reprinted, so the copy you have purchased may already incorporate some or all of these corrections.

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<th>BOOK PAGE</th>
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| Page 558  | Question 16, answer choice (D) should read:  
(D) 1s^22s^22p^2  
Question 17, answer choices (C) and (D) should read:  
(C) Propane  
(D) Propene |
| Page 567  | Question 12. The quoted passages in answer choices (B) and (C) should be switched:  
(B) "One of them showed that corticosterone..."  
(C) "Researchers read these results..." |
| Page 603  | Answer explanation 17, sentences 3 and 4 should read: Since there are three carbon atoms, the prefix of the molecule name is prop. Therefore, the name given to the molecule is propene. |
| Page 629  | The prompt should read: Discuss a solution to the problem of an increasingly partisan, polarized, and unproductive Congress. |
| Page 643  | Question 9, the answer choices should read:  
(A) 0.51 atm  
(B) 1.27 atm  
(C) 4.42 atm  
(D) 11.3 atm |
| Page 692  | The first section of answer explanation 9 (up to "Choice (A) is incorrect...") should read: **The correct answer is (C).** The osmotic pressure pushes solvent from a more dilute area to a more concentrated area in an attempt to achieve equilibrium between the two areas. Osmotic pressure, \( \pi \), can be calculated by the equation \( \pi = MRT \) where \( M \) = molarity of the solution (mole/L), \( R \) is the ideal gas constant.
rate constant (0.082 atm L/M K), and T is temperature in Kelvin K (310). Moles of Na + M is calculated to be 0.17 Mole /L from the following: 1.0 g Na+ divided by 0.25 L, then divided by 23 (molecular wt of Na+).